Science Knowledge Organiser Glossopdale

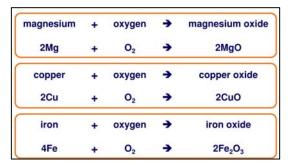
Y7 Reactions: Metals and non-metals

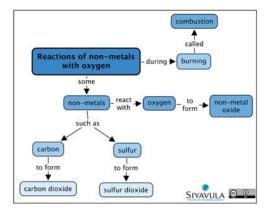
Reactions of metals with acid The general equation for the reaction of a metal with acid is:

Metal + acid → salt + hydrogen

The test for hydrogen is sometimes called the squeaky pop test. Hydrogen makes a 'squeaky pop' when it is placed near a lit wooden splint.

Metals can be arranged as a **reactivity series** in order of how readily they react with other substances.





How to reme	ember the Reactiv	vity Series?
Please	Potassium	Most reactive
Stop	Sodium	
Calling	Calcium	
Me	M agnesium	
Α	A luminium	
Careless	(Carbon)	
Z ebra	Zinc	
Instead	Iron	
Try	Tin	
Learning	Lead	
How	(Hydrogen)	
Copper	Copper	
Saves	Silver	
Gold	Gold	Least reactive

Key words		
Displacement	Reaction where a more reactive metal takes the place of a less reactive metal in a compound.	
Metals	Shiny, good conductors of electricity and heat, malleable and ductile, and usually solid at room temperature.	
Non-metals	Dull, poor conductors of electricity and heat, brittle and usually solid or gaseous at room temperature.	
Oxidation	Reaction in which a substance combines with oxygen.	
Reactivity	The tendency of a substance to undergo a chemical reaction.	

Properties of metals and non metals			
Property	Metals	Non-metals	
Appearance	Shiny	Dull	
State at room temperature	Solid (except mercury, which is a liquid)	About half are solids, about half are gases, and one (bromine) is a liquid	
Density	High (they feel heavy for their size)	Low (they feel light for their size)	
Strength	Strong	Weak	
Malleable or brittle	Malleable (they bend without breaking)	Brittle (they break or shatter when hammered)	
Conduction of heat	Good	Poor (they are insulators)	
Conduction of electricity	Good	Poor (they are insulators, apart from graphite)	
Magnetic material	Only iron, cobalt and nickel	None	
Sound when hit	They make a ringing sound (they are sonorous)	They make a dull sound	